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Prediction of chemical shifts for protein (mis-)folding studies

1. Introduction

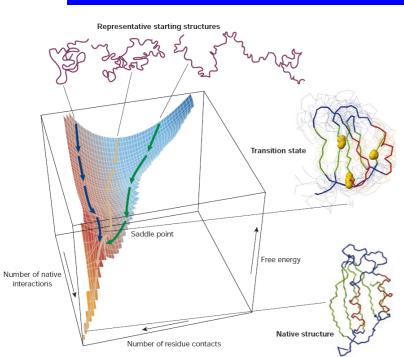


Fig. 1: Schematic representation of a simple energy landscape. Different starting structures of the same protein travel along the landscape towards their native structure, possibly via intermediates, or transition states. (from [2])

acid chains that are formed at the ribosome. While initially unfolded, the chains usually have to fold into a specific three-dimensional 'native' structure in order to become biologically active [1]. This folding event, leading from a state of high energy to a minimum energy state, can be described as a pathway on a 'free energy landscape'

(figure 1): like a skier the protein tries to get from a mountain peak into the valley without having to put in additional energy. In some cases there might be several favourable pathways to do so. However, especially when with age the body's quality control mechanisms start to weaken, proteins might choose an incorrect folding pathway, mis-fold, and form aggregates: they end up in the wrong valley from where there is no return.

Such aggregates are believed to trigger diseases like Alzheimer's and late onset diabetes [3].

can help defining energy
landscapes and further our understanding of mis-folding events [4].
The behaviour of very small molecules can be treated via quantum mechanical calculations, but for more complex proteins it can only be approximated, for example via Molecular Dynamics (MD) computer

simulations (figure 2). These methods rely on the laws of Physics and information obtained from experiments to approximate atomic movements.

We try to increase accuracy and speed of MD simulations by using experimental data from Nuclear Magnetic

In particular, we presently try predicting a property called 'chemical shifts' (c.s.) which is a very sensitive measure of an atom's atomic environment [7]. By lefining penalty functions that depend on the difference between calculated c.s. of a simulated protein structure and experimental c.s of a target structure we hope to be able to guide computer-based folding towards

that target structure.

2. Research
Project

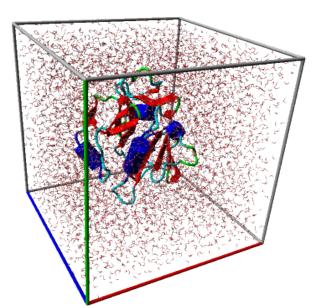


Fig. 2: A Molecular Dynamics simulation box containing the protein dihydrofolate reductase and water molecules as solvent (from [5]). The hundreds of millions of possible pair-wise interactions and resulting movements in such complex systems can only be approximated.

4. Project Status

Based on data fitting between chemical shift data [13, 14] and protein structure [15] a preliminary version of the prediction algorithm has been developed. The main 'CamShift' equation consists of a linear combination of terms:

$$\delta = \delta_{c} + \delta_{dbb} + \delta_{dsc} + \delta_{dcb} + \delta_{hb} + \delta_{ar}$$

where the scalar δ on the left hand side is the chemical shift value for a given atom and the terms on the right depend on inter-atomic distances (δ_{dbb} , δ_{dsc} , δ_{dcb}), as well as capturing more complex hydrogen bonding and aromatic ring effects (δ_{hb} , δ_{ar}). δ_c is a constant.

At present, this equation was trained with data for two different atom types in a protein: $H\alpha$ and $C\alpha$.

6. Future

Work

3. Aim

A number of chemical shift predictors exist which use a variety of techniques from artificial neural networks to protein homology [8-11]. A recent study using Random Forests non-linear regression suggest that there is room for improvement in all cases [12].

Our aim is twofold:

- Achieve better performance than current approaches
- Develop a fast-to-compute, easily differentiable function that, unlike existing predictors, can easily be implemented into MD simulations

5. Preliminary Results

This first version of the predictor achieves a root mean square deviation (rmsd) between predicted and experimental chemical shifts of 0.28 ppm¹ for H α and 1.52 ppm for C α , which is a significant improvement to the results for an uninformed best guess (based on always predicting the mean of the sample) resulting in rmsds of 0.57 ppm for H α and 4.9 ppm for C α .

However, this is not yet as good as other predictors that report accuracies of up to 0.23 ppm (H α) and 0.98 ppm (C α) in some cases [11].

In short, future steps will be:

- Optimization and extension of terms for c.s. contributions
- Extension of the algorithm to work on other atom types
- Detailed comparison of predictive performance with that of existing predictors
- Implementation of CamShift into Molecular Dynamics package
 - Applying new implementation to protein folding simulations

Acknowledge ments

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¹ppm = parts per million, the unit of chemical shifts

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The foreground graphic shows the side-chain of the natural amino acid tyrosine